

**M.Sc.-Ist Semester (Biotechnology & Microbiology)**

COURSE CODE (CREDITS):20MSBT111 (3)

MAX. MARKS: 15

COURSE NAME: **Biochemistry**

COURSE INSTRUCTORS: **Dr. Jitendraa Vashistt**

MAX. TIME: 1 Hour

*Note: (a) All questions are compulsory. (b) Marks are indicated against each question in square brackets. (c) The candidate is allowed to make Suitable numeric assumptions wherever required for solving problems.*

**Q1.** How do you define a 'buffer' solution? What is the significance of a buffer in biological experiments? Justify your answer with suitable example. **(3 marks)**

**Q2.** How do you calculate and prepare the following. **(3 marks)**

a) A 100 ml solution of 100mM NaOH.

b)  $H^+$  in a solution of 100mM NaOH.

(Given Molecular weight of NaOH = 40 g/mol)

**Q3.** Explain a classical experiment which laid a foundation of chemical basis of origin of life that simulated the conditions present in the atmosphere of the early, prebiotic earth. **(3 marks)**

**Q4.** DNA has an overall negative charge and it needs to be placed in the tight packing in the nucleus of a eukaryotic cell. Which biomolecule and its basic units are usually present in packing of DNA in the nucleus? Justify your answer. **(3 marks)**

**Q5.** Define the following in brief. **(3 marks)**

a) Cohesive and adhesive interaction with respect to water

b) Hydrogen bonding and stability of a biomolecule