

JAYPEE UNIVERSITY OF INFORMATION TECHNOLOGY, WAKNAGHAT
TEST -3 EXAMINATIONS- 2023

M.Sc.-Ist Semester (BT)

COURSE CODE (CREDITS):20MS1BT111, 03

MAX. MARKS: 35

COURSE NAME: **Biochemistry**

COURSE INSTRUCTOR: **Dr. Jitendraa Vashistt**

MAX. TIME: 2 Hour

Note: (a) All questions are compulsory. (b) Marks are indicated against each question in brackets. (c) The candidate is allowed to make Suitable numeric assumptions wherever required for solving problems

- Q1a).** Can it be possible that cytosine nucleotide convert into Uracil nucleotide? Justify your answer with suitable structural elucidations? **(3 marks)**
- Q1b).** How do you classify DNA in terms of its polarity; hydrophilic or hydrophobic? Justify your answer with suitable structural elucidations? **(3 marks)**
- Q2.** Why there is a huge market of Biotin supplements? Define the significance of this molecule in human metabolism. **(3 marks)**
- Q3.** Some people do dieting to control their overweight, with a common practice by avoiding all carbohydrates in their diet. Is it a correct practice? Justify your answer. **(3 marks)**
- Q4.** There is a common practice in houses as well as in restaurants that cooking oil used repeatedly by reheating. What is the major ill-effect by reheating any cooking oil in terms of the structural changes? **(3 marks)**
- Q5a).** Calculate the number of acetyl COA and ATPs generated by a fatty acid in which length of carbon atoms is 20 (C20) and define the role of carnitine shuttle in fatty acid metabolism. **(3 marks)**
- Q5b).** How do you calculate the pH of a Sodium acetate buffer, if sodium acetate and acetic acid are equimolar and pKa is 4.75? **(3 marks)**
- Q6.** How two biochemical processes 'glycolysis' and 'gluconeogenesis' are interlinked with each? Also explain the three crucial steps and their alternate reactions by which glucose gets synthesized back from pyruvate **(4 marks)**
- Q7:** How ammonia is produced in humans and why it is toxic to human body? Explain the metabolic process by which detoxification of ammonia occurs in liver. **(5 marks)**
- Q8.** Calculate the Michaelis-Menten constant (MM) for an enzymatic reaction which has K_1 constant of Enzyme+ Substrate to form Enzyme: substrate is $1 \times 10^9 \text{M}^{-1} \text{sec}^{-1}$, K_{-1} constant of reverse reaction is $1 \times 10^6 \text{M}^{-1} \text{sec}^{-1}$. When the reaction proceeds to form product and dissociation of enzyme from substrate occur, the K_2 constant of Enzyme + Product is $1 \times 10^9 \text{M}^{-1} \text{sec}^{-1}$. **(5 marks)**